Synthesis and characterization of ZnO nano particles using Gigantic Swallow Wort leaves using by green chemical reduction method

Mr. Pankaj Shende¹, Syed Tanveer² and A. Nagesh³

Lecturer in Department of Mineral Processing Engineering, EIT, MAI-NEFHI, ASMARA, ERITREA Post Box No. 12676 Asst.Prof. of physics, Brindavan Institute of Technology & Science, Peddatekur (V), Kurnool 518218 Asst.Prof. of physics, EIT, MAI-NEFHI, ASMARA, ERITREA, Post Box No. 12676.

Abstract

Present study focuses on the extraction of ZnO nanoparticles by Green chemical reduction method from the bio components of leaves extract of Gigantic Swallow Wort. X-Ray Diffraction (XRD), FT-IR Spectroscopy characterizations was done for synthesized ZnO nanoparticles. X-ray diffraction studies showed that the particles are hexagonal in nature

Keywords-FTIR, Green Chemical reduction method. XRD

1. INTRODUCTION

Semiconductors with dimensions in the nanometer realm are important because their electrical, ical and chemical properties can be tuned by changing the size of particles. But Zinc oxide (ZnO) is a wide band gap (3.37 eV) and high excitation binding energy of (60 meV) at room temperature [1, 2] and has unique optical and as well as excellent thermal and chemical stability [3]. ZnO nanoparticles have very large surface area with potentially low manufacturing cost. The shape and size of nano materials immensely affect their physical and chemical properties. Among the metal oxide nanoparticles, zinc oxide is interesting because it has vast applications in various areas such as optical, piezoelectric, magnetic, and gas sensing. Besides these properties, ZnO nanostructure exhibits high catalytic efficiency, strong adsorption ability and are used more and more frequently in the manufacture of sunscreens [4], ceramics and rubber processing, wastewater treatment, and as a fungicide [5, 6]. In fact, n-ZnO usage may overtake nano-titanium dioxide (nTiO₂) in the near future as it can absorb both UV-A and UV-B radiation while nTiO₂ can only block UV-B, and

thereby offering better protection and improved opaqueness [5]. Several physical and chemical procedures have been used for the synthesis of large quantities of metal nanoparticles in relatively short period of time. Chemical methods lead to the presence of some toxic chemicals adsorbed on the surface that may have adverse effects in medical application [7]. Currently, plant-mediated biological synthesis of nanoparticles is gaining importance due to its simplicity, eco-friendliness and extensive antimicrobial activity [8, 9]. It has wide applications in different industries including photodetectors[10], sensors [11], solar cells [12], antibacterial or medical products [13-15], cosmetics [16], etc. ZnO nanoparticles can be synthesized by various chemical for physical methods such as precipitation [17], sol-gel [18], solvo/hydrothermal [19], chemical vapor deposition [20], spray pyrolysis [21], etc. Biosynthesis of zinc oxide nanoparticles by plants such as Aloe vera [22] and gold nanoparticles by alfalfa [23, 24], Cinnamomum camphora [25], neem [26], Emblica officianalis [27], lemongrass [28] and tamarind [29] have been reported. To the best of our knowledge, the present study is the first report on ZnO nanoparticles synthesized using Gigantic Swallow Wort leaves

2. MATERIALS AND METHODS

A. Materials

Zinc nitrate and utilized ingredients with analytical grade chemicals were purchased from Merck and used without further purification. Distilled and deionized water was used in this work. The leaves of Gigantic Swallow Wort (Figure 1) plant collected from our surround places Zinc nitrate is a crystalline and inorganic chemical compound with the formula Zn (NO_3)₂. Zinc Nitrate is soluble in both alcohol and water. The structure of zinc nitrate is shown in figure 1.

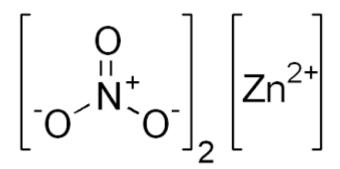


Figure 1 structure of zinc nitrate

B. Preparation of leaf extract

For the preparation of leaves extract fresh leaves were collected from Gigantic Swallow Wort plants (figure 2). The leaves were washed several times with water to remove the dust particles and then sun dried to remove the residual moisture.



Figure 2 Gigantic Swallow Wort plant

The dried leaves were cutted and grinded for powder. Then taking 20 gm of dried Gigantic Swallow Wort leaves boiled in 250 ml of deionised water for one hour at 70-80 $^{\circ}$ C. The mixture solution cooled at room temperature. The leaves extract (Figure 3) filtered by using watts man filter paper.



Figure 3. Gigantic Swallow Wort plant leaf extract

C. Green synthesis of ZnO nanoparticles using leaf Extract of Giganticon SwallowWort Plant.

For the ZnO nanoparticles synthesis, 100 ml of GiganticSwallowWort leaf extract was taken boiled to 60-70 ^oC. using magnetic stirrer heater. 5 grams of Zinc Nitrate was

added to the solution. This mixture is then boiled until it reduced to a deep yellow coloured paste. This paste was then collected in a ceramic crucible and heated in an air heated furnace at 300 degree Celsius for 2 hours. A light yellow coloured powder was obtained and this was carefully collected and packed for characterization purposes.

3. RESULTS AND DISCSSION

A.FTIR Spectra analysis

The FTIR spectrum of ZnO nanoparticles is shown in Figure 4. The IR spectrum of transmittance was taken by using a Bruker FT-IR instrument operating at a resolution of 2000-400 cm⁻¹. In IR spectra, the absorption peak at 473.33 cm⁻¹ indicates the presence of ZnO nanoparticles. The sharp characteristic peaks are also observed in FTIR spectrum of ZnO nanoparticles synthesized from Zinc Nitrate suggesting the high crystalline nature of ZnO nanoparticles

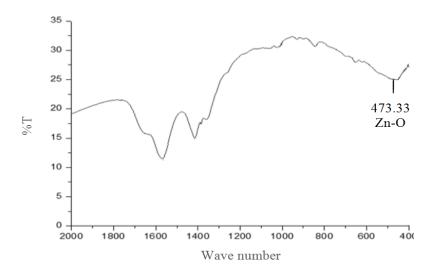


Figure 4. FTIR spectrum of ZnO nanoparticles

B. X-ray diffraction analysis of ZnO nanoparticles.

XRD spectrum of the prepared ZnO nanoparticles was carried out using XRD for 20 values ranging from 20 to 70° using CuK α radiation at $\lambda = 1.5406$ Å. In ZnO, the 20 values with (hkl) plane at 33.5° (100), 35.2° (002), 37.1° (101), 47.4° (102), 57.5° (110), 62.8° (103) and 67.8° (201) were observed. The spectrum (Figure 5.) confirmed the hexagonal zinc oxide structure for ZnO nanoparticles. The average particle size (D) of synthesized nanoparticles was calculated using the well known Scherrer formula

104

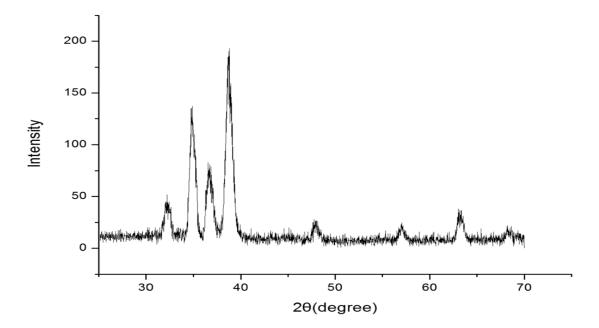


Figure 2 X-Ray diffraction spectrum of ZnO nanoparticles

4. CONCLUSION

ZnO NPs was synthesized by the green chemical reduction method using Gigantic-Swallow-Wort leafextract is simple and cost effective. The as prepared ZnO nanoparticles were characterized using several techniques such as XRD and FTIR. The FT-IR studies showed an absorption peak at 473.33 cm⁻¹ (Zn-O linkage) which indicated the formation of zinc oxide nanoparticles. From XRD analysis, the structure of the ZnO particles is confirmed as hexagonal with average particle size 28.38 nm.

REFERENCES

- [1]. E. Roduner, Size matters: why nanomaterials are different. Chemical Society Reviews 35 (7), (2006), 583-592.
- [2] M.R Jones et al., Templated Techniques for the Synthesis and Assembly of PlasmonicNanostructures., Chemical Reviews, 111 (6), (2011), 3736-3827.
- [3] S.Iravani, Green synthesis of metal nanoparticles using plants. Green Chemistry. 13 (10), (2013), 2638-2650
- [4] R. Seshadri, in: C.N.R. Rao, A. Muller, A.K. Cheetham (Eds.), "The Chemistry of Nanomaterials, Vol.1, Wiley-VCH VerlagGmbH, Weinheim, 2004, pp. 94-112.
- [5] L. Theodore, Nanotechnology: Basic Calculations for Engineersand Scientists, Wiley, Hoboken, 2006.

- [6] X. Wang, J. Lu, M. Xu, B. Xing, Sorption of pyrene by regularand nanoscaled metal oxide particles: influence of adsorbedorganic matter, Environmental Science and Technology 42 (2008) 7267-7272.
- [7] D. Jain, H.K. Daima, S. Kachhwaha, S.L. Kothari, Synthesis ofplant-mediated silver nanoparticles using papaya fruit extract andevaluation of their antimicrobial activities, Digest Journal of Nanomaterials and Biostructures 4 (2009) 557-563.
- [8] A. Saxena, R.M. Tripathi, R.P. Singh, Biological synthesis of silvernanoparticles by using onion (Allium cepa) extract and their antibacterial activity, Digest Journal of Nanomaterials and Biostructures5 (2010) 427-432.
- [9] N. Khandelwal, A. Singh, D. Jain, M.K. Upadhyay, H.N.Verma, Green synthesis of silver nanoparticles using Argimonemexicana leaf extract and evaluation of their antimicrobialactivities, Digest Journal of Nanomaterials and Biostructures5 (2010) 483-489.
- [10] Liang S, Sheng H, Liu Y, Huo Z, Lu Y, Shen H: "ZnO Schottky ultravioletphotodetectors". J Cryst Growth 2001, 225 (no. 2-4):110-113.
- [11] Lin Y, Zhang Z, Tang Z, Yuan F, Li J: Characterisation of ZnObasedVaristors Prepared from Nanometre Precursor Powders. Adv. Mater. Opt.Electron 1999, 9:205-209.
- [12] Golego N, Studenikin SA, Cocivera M: "Sensor Photoresponse of Thin-FilmOxides of Zinc and Titanium to Oxygen Gas". J Electrochem Soc 2000, 147:1592.
- [13] Sawai J: Quantitative evaluation of antibacterial activities of metallicoxide powders (ZnO, MgO and CaO) by conductimetric assay. J MicrobiolMethods 2003, 54 (2):177-182.
- [14] Zhang L, Jiang Y, Ding Y, Povey M, York D: Investigation into theantibacterial behaviour of suspensions of ZnO nanoparticles (ZnOnanofluids). Journal of Nanoparticle Research 2006, 9 (3):479-489.
- [15] Raghupathi KR, Koodali RT, Manna AC: Size-dependent bacterial growthinhibition and mechanism of antibacterial activity of zinc oxidenanoparticles. ACS 2011, 27 (7):4020-4028.
- [16] Vaezi MR, Sadrnezhaad SK: Nanopowder synthesis of zinc oxide viasolochemical processing. Mat Des 2007, 28 (2):515-519.
- [17] Wang L, Muhammed M: Synthesis of zinc oxide nanoparticles with controlled morphology". J Mater Chem 1999, 9:2871-2878.
- [18] Yan C. Chen Z. Zhao X: Enhanced electroluminescence of ZnOnanocrystalline annealing from mesoporous Solid precursors. StateCommun 2006, 140 (1):18-22.
- [19] Pan A, Yu R, Xie S, Zhang Z, Jin C, Zou B: ZnO flowers made up of thinnanosheets and their optical properties. J Cryst Growth 2005, 282 (1-2):165-172.
- [20] Wu J-J, Liu S-C: Catalyst-Free Growth and Characterization of ZnONanorods. J Phys Chem B 2002, 106 (37):9546-9551.
- [21] Ghaffarian H, Saiedi M: "Synthesis of ZnO Nanoparticles by Spray PyrolysisMethod". Iran J Chem Chem Eng 2011, 30 (no. 1):1-6.

- [22] G. Sangeetha, S. Rajeshwari, R. Venckatesh, Green synthesis ofzinc oxide nanoparticles by aloe barbadensis miller leaf extract:structure and optical properties, Materials Research Bulletin 46 (2011) 2560-2566.
- [23] J.L. Gardea-Torresdey, J.G. Parsons, E. Gomez, J. PeraltVidea, Formation and growth of Au nanoparticles inside livealfalfa plants, Nanoletters 2 (2002) 397-401.
- [24] J.L. Gardea-Torresdey, E. Gomez Jr., Peralta-VideaJr, J.G.Parsons, H. TroianiM. Jose-Yacaman, Alfalfa Sprouts: ANatural Source for the Synthesis of Silver Nanoparticles, Langmuir 19 (2003) 1357-1361.
- [25] J. Huang, Q. Li, D. Sun, Y. Lu, Y. Su, X. Yang, Biosynthesis ofsilver and gold nanoparticles by novel sundried Cinnamomumcamphora leaf, Nanotechnology 18 (2007) 105104-105114.
- [26] S. Shiv Shankar, A. Rai, A. Ahmad, M. Sastry, Rapid synthesis of Au, Ag, and bimetallic Au core-Ag shell nanoparticles using neem (Azadirachta indica) leaf broth, Journal of Colloid and InterfaceScience 275 (2004) 496-502.
- [27] B. Ankamwar, D. Chinmay, A. Absar, S. Murali, Biosynthesisof gold and silver nanoparticles using emblica officinalis fruitextract, their phase transfer and transmetallation in an organicsolution, Journal of Nanoscience and Nanotechnology 10 (2005)1665-1671.
- [28] S.ShivShankar, A.Rai, A.Ahmad, M.Sastry, RapidsynthesisofAu, Ag, andbimetallicAucore-Agshellnanoparticlesusingneem (Azadirachtaindica)leafbroth, JournalofColloidandInterfaceScience275 (2004)496-502.
- [29] B. Ankamwar, M. Chaudhary, Gold nanotriangles biologicallysynthesized using tamarind leaf extract and potential applicationin vapor sensing, Synthesis and Reactivity in Inorganic andMetal-Organic Chemistry 35 (2005) 19-26.