Novel Synthesis of Heterocyclic Substituted 1,5-Benzothiazepines

Parthasarathy Muthuraman, Sreeraj Gopi and S. Narasimhan*

Asthagiri Herbal Research Foundation, 162-A, Perangudi, Industrial Estate, Chennai-600092, India E-mail: narasihan-s@yahoo.com

Abstract

Owing to their well known bioactivities, the synthesis and chemical transformations of various groups of benzothiazepines have been studied and the procedures used have also been summarized. , we have developed a new methodologies for the synthesis of 1,5-benzothiazepines by microwave irradiation. Our method has many advantages over existing methods, including moderate yield, simple work-up, shorter reaction span, no side reactions no critical purification method. This procedure represents a convenient, economic and environmentally friendly process for the synthesis of 1,5-benzothiazepines were prepared by the reaction of substituted sufanyl derivatives with 2-amino phenol. The structures of the synthesised products were confirmed by NMR.

Keywords: Benzothiazepines, Bioactivities, Pharamacophore, 2-amino phenol

Introduction

Benzothiazepine and its derivatives show a wide spectrum of pharmacological activities such as coronary vasodilatory, tranquilizer, antidepressant, CNS stimulant, antihypertensive, calcium channel blocker, antiulcer, calcium antagonist and antimicrobial agents.10,11 Benzothiazepine derivatives have been reported to be more potent selective inhibitors of the mitochondrial Na+–Ca2+ exchangers.Several naturally occurring and synthetic benzofuran derivatives are known to be associated with biological and pharmacological activities such as anti-inflammatory13, anti-implantation14, anticancer15-17, pesticidal18, nematicidal19, antiuterotropic20 activities. The applications of benzofuran derivatives were also extended to textile and paper industries. The 1,5-benzothiazepine moiety is a privileged class of

pharamacophore, as compounds bearing this structural unit possess a broad spectrum of biological activities. The high biological activity of these molecules prompted us to study the synthesis of novel molecules of these class. In the present study, we report the synthesis of novel benzothiazepines substituted with hetrocycles in Green chemical approach.

The overall scheme is given below.



Materials

Reaction vessels and other glass equipment used for the experiments were thoroughly dried in an oven and cooled under a stream of dry nitrogen just prior to use. All reagents, solvents were purchased from Sigma-Aldrich and were used as supplied. Thin-layer chromatography (TLC) was performed on 0.25 mm pre coated silica gel 60 F254 aluminum sheets and column chromatography on silica gel 60 (0.063-0.2 mm) as well as silica gel 60 (<0.063 mm), products of Merck & Co. (Darmstadt, Germany).

Experimental

General

All bp's and mp's were uncorrected. The IR spectra were scanned with a Perkin-Elmer 783 spectrophotometer and only the pertinent values were expressed, in cm⁻¹. The C¹³ & H¹NMR spectra were recorded, with a Bruker (500 MHz) spectrometer, with TMS as internal standard.CDCl₃ was used as the solvent. The chemicalshift (d) and

coupling constant (J) values were expressed in ppm and Hz only. The GLC analyses were carried out on a Shimadzu GC-7A chromatograph fitted with a flame ionization detector and glass packed column for routine analysis and a capillary column for the determination of isomeric compositions. The mass spectra (EI) was recorded at 70 eV with a Shimadzu GC-MS QP-1000A spectrometer. Unless otherwise mentioned, the organic extracts were dried over anhydrous Na₂SO₄.

Synthesis of Sufanyl Derivative of Acetophenone

Sodium tert-butaoxide (10mmol) was taken in a dried round bottom flask with side arm adaptor and septum. The flask was purged three times with Nitrogen and subsequently immersed into an ice bath (0 °C). 100ml of dry THF was added and stirred. Acetophenone was added and followed by CS₂, then MeI₂ added and allowed to stirred at rt for 16 hrs. The reaction was monitored by TLC. After the completion of the reaction, the solvents were distilled out and the product obtained as crystalline solid. The melting point was determined , which was matching with the literature value.



Synthesis of Substituted Benzothiazepines

The substituted sulfanyls (10mmol) were taken in a dried round bottom flask with side arm adaptor and septum. The flask was purged with nitrogen and connected to a mercury bubbler and 10g of dry silica gel was added to the flask. 2-amino benzothiol was added through the side arm, using a solid addition funnel. Both reactant was taken in same equalent. The reaction mixture was kept under microwave irradiation, at 70°C for half an hour. The reaction was further monitored with TLC and after completion of reaction, crude product extracted in chloroform. The organic layer was separated and the aqueous layer extracted two times with ethyl acetate. The organic portions combined and dried with anhydrous sodium sulphate. The product was concentrated by rotary evaporator and purified by column chromatography. The product was characterized by NMR, MS etc



Synthesised Compounds





 ¹³C (CDCl3)- 164.9,152.9,127.9, 122.3, 115.9,129.1,128.9,116.3,21.8,9.9
¹H -7.57, 7.20, 6.57, 5.2, 6.49, 2.09





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Result and Discussion

In summary, we have developed a new methodologies for the synthesis of 1,5benzothiazepines by microwave irradiation. We have successfully substituted few hetrocyclic moiety in the 1,5-benzothiazepines. Our method has many advantages over existing methods, including moderate yield, simple work-up, shorter reaction span, no side reactions no critical purification method. This procedure represents a convenient, economic and environmentally friendly process for the synthesis of 1,5benzothiazepines.

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